

**GUJARAT TECHNOLOGICAL UNIVERSITY**

**BE- SEMESTER-III EXAMINATION – WINTER 2025**

**Subject Code:3131904**

**Date:12-12-2025**

**Subject Name: Material Science and Metallurgy**

**Time:10:30 AM TO 01:00 PM**

**Total Marks:70**

**Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.
4. Simple and non-programmable scientific calculators are allowed.

**Marks**

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|------------|--|-----------|
| <b>Q.1</b> | <b>(a)</b> Give difference between Metals & Non-metals.  | <b>03</b> |
|            | <b>(b)</b> What is Unit cell? Explain different types of unit cell with neat sketch.   | <b>04</b> |
|            | <b>(c)</b> Draw and explain iron carbon diagram.   | <b>07</b> |
| <b>Q.2</b> | <b>(a)</b> Enlist and Explain important properties of materials.   | <b>03</b> |
|            | <b>(b)</b> Explain Recovery, recrystallization, and grain growth.  | <b>04</b> |
|            | <b>(c)</b> Explain and differentiate Edge dislocation and Screw dislocation with neat sketch.  | <b>07</b> |
|            | <b>OR</b>  |           |
|            | <b>(c)</b> Enlist and explain all point defects.   | <b>07</b> |
| <b>Q.3</b> | <b>(a)</b> Difference between micro and macro examination.   | <b>03</b> |
|            | <b>(b)</b> Briefly explain Gibb's Phase rule.  | <b>04</b> |
|            | <b>(c)</b> What is a substitutional solid solution? Explain the Hume Rothery rule for the formation of a substitutional solid solution.              | <b>07</b> |
|            | <b>OR</b>  |           |
| <b>Q.3</b> | <b>(a)</b> Discuss merits and de-merit of powder metallurgy process.   | <b>03</b> |
|            | <b>(b)</b> Explain Jominy end quench test.   | <b>04</b> |
|            | <b>(c)</b> What is Powder Metallurgy? Explain the process steps of Powder Metallurgy.  | <b>07</b> |
| <b>Q.4</b> | <b>(a)</b> Draw and explain Pearlite.  | <b>03</b> |
|            | <b>(b)</b> What is allotropic form ? explain different allotropic form of iron.  | <b>04</b> |
|            | <b>(c)</b> Explain TTT diagram in detail with sketch.  | <b>07</b> |
|            | <b>OR</b>  |           |
| <b>Q.4</b> | <b>(a)</b> What is eutectic point ? Explain in brief.  | <b>03</b> |
|            | <b>(b)</b> Briefly explain Gibb's Phase rule.  | <b>04</b> |
|            | <b>(c)</b> Draw and explain the phase diagram of binary system having complete soluble in liquid but total insoluble in solid phase with lever rule. | <b>07</b> |
| <b>Q.5</b> | <b>(a)</b> Explain Optical Metallurgical Microscope with neat sketch.  | <b>03</b> |
|            | <b>(b)</b> Explain the steps of Dye Penetration Testing with neat sketch.  | <b>04</b> |
|            | <b>(c)</b> Explain the principle of Radiography testing. With its use in metal testing.  | <b>07</b> |
|            | <b>OR</b>  |           |
| <b>Q.5</b> | <b>(a)</b> What is heat treatment ?  | <b>03</b> |
|            | <b>(b)</b> Differentiate Annealing Vs Normalising.   | <b>04</b> |
|            | <b>(c)</b> Explain Flame Hardening Process with neat sketch.   | <b>07</b> |

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