

GUJARAT TECHNOLOGICAL UNIVERSITY**BE- SEMESTER-VII (NEW) EXAMINATION – WINTER 2024****Subject Code:3171924****Date:16-12-2024****Subject Name: Principles of Combustion****Time:10:30 AM TO 01:00 PM****Total Marks:70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.
4. Simple and non-programmable scientific calculators are allowed.

MARKS

Q.1	(a)	Explain heterogeneous reactions.	03
	(b)	Explain following terms: (1) Equivalence Ratio (2) Stoichiometry	04
	(c)	Explain the H ₂ -O ₂ system in details	07
Q.2	(a)	Enlist various factors affecting the flame length.	03
	(b)	List any four Applications of turbulent flames	04
	(c)	Explain Well Stirred flow reactor and Derive expression for it.	07
OR			
	(c)	Write short note on Arrhenius rate expression.	07
Q.3	(a)	Explain heterogeneous reactions	03
	(b)	Determine the stoichiometric equation for combustion of (a) Oxygen (b) Air.	04
	(c)	Write short note on rate of reaction and their temperature dependence.	07
OR			
Q.3	(a)	Explain Mixture fraction in brief.	03
	(b)	Explain counter flow flames in details.	04
	(c)	Write short note on Pseudo-first-order reactions and the “fall-off” range	07
Q.4	(a)	Explain Soot Formation	03
	(b)	Explain wrinkled laminar flame regime with neat sketch	04
	(c)	Write short note on partial equilibrium assumption.	07
OR			
Q.4	(a)	Explain jet flames in short.	03
	(b)	What are the objectives of turbulent diffusion flame analysis?	04
	(c)	Explain the fundamental one dimensional conservation laws in details	07
Q.5	(a)	Discuss the concept of Adiabatic Flame Temperature.	03
	(b)	Differentiate homogeneous and heterogeneous reaction with suitable example.	04
	(c)	Explain laminar flame structure with figure.	07
OR			
Q.5	(a)	Explain the Burning of Carbon with chemical equations	03
	(b)	Explain conserved scalar concept in short.	04
	(c)	Explain wrinkled laminar flame regime with neat sketch.	07
