

GUJARAT TECHNOLOGICAL UNIVERSITY**BE - SEMESTER-VI (NEW) EXAMINATION – SUMMER 2024****Subject Code:3160501****Date:15-05-2024****Subject Name:Mass Transfer Operations II****Time:10:30 AM TO 01:00 PM****Total Marks:70****Instructions:**

1. Attempt all questions.
2. Make suitable assumptions wherever necessary.
3. Figures to the right indicate full marks.
4. Simple and non-programmable scientific calculators are allowed.

- | | Marks |
|--|--------------|
| Q.1 (a) Define: 1) Absolute Humidity 2) Unbound Moisture 3) Humid Volume | 03 |
| (b) Generally, adsorption is preferable at low temperature and high pressure. True or false. Justify. | 04 |
| (c) Enlist various types of distillation in brief and explain azeotropic distillation in detail. | 07 |
| Q.2 (a) Classify drying equipment in depth. | 03 |
| (b) Explain q line for distillation. Also derive equation for q-line. | 04 |
| (c) A fractional column separates a liquid mixture entering at 5000 kmol/h containing 50 mole % A and rest B into an overhead product of 95 mole % A and a bottom product of 96 % mole B. $R = 2R_{\min}$ is used and feed enters at boiling point. Determine number of theoretical stages required and feed point location. | 07 |

x	0.03	0.06	0.11	0.14	0.26	0.39	0.53	0.66	0.76	0.86	1
y	0.08	0.16	0.27	0.33	0.50	0.63	0.71	0.83	0.88	0.93	1

OR

- (c) It is desired to separate a mixture of 50 % vapor and 50 % liquid in a plate type distillation column. The feed contains 45 mole % A and top product has 96 mole % A whereas bottom product has 5 mole % A. Determine minimum reflux ratio and number of theoretical plates needed if reflux ratio is twice the minimum is used.

x	0	0.1	0.16	0.3	0.4	0.5	0.6	0.7	0.8	0.9	1
y	0	0.215	0.30	0.52	0.625	0.725	0.78	0.89	0.89	0.95	1

- Q.3** (a) Steam distillation takes place at reduced temperature. Justify. Also explain the reason behind the same. **03**
- (b) Classify cooling towers in detail. **04**
- (c) Explain principle and working of fluidized bed dryer with neat sketch. **07**

OR

- Q.3** (a) Briefly explain steam distillation by stating proper example. **03**
- (b) Distillation tower can not be operated at minimum reflux or total reflux. Justify. **04**
- (c) Derive equation for retention time in continuous dryer at low temperature. **07**
- Q.4** (a) Discuss about the different factors affecting drying operation. **03**

- (b) Explain equilibrium with reference to adsorption. State characteristics of adsorbent as well. **04**
- (c) A woolen cloth is dried in hot air dryer from 100 % to 10 % moisture. If the equilibrium and critical moisture content is 6 % and 55 % respectively. Estimate the reduction in time needed to dry the solid to 16 % moisture instead of 10 % under the same conditions. (All the moisture contents are on dry basis) **07**
Also mention major applications of drying and resistances acting while drying.

OR

- Q.4** (a) Define: 1) Relative humidity, 2) Humid heat, 3) Dew point **03**
- (b) Explain principles of ion exchange in brief. Also list its applications. **04**
- (c) Explain the theory of adiabatic saturation temperature and derive the equation for adiabatic saturation temperature. **07**

- Q.5** (a) Brief about optimum reflux ratio. **03**
- (b) Explain moisture movement mechanism within solid for drying. **04**
- (c) Explain Freundlich equation. Derive the relation for two stage counter current adsorption using the Freundlich equation. **07**

OR

- Q.5** (a) Brief about elution and chromatography. **03**
- (b) Explain range and approach with reference to cooling tower. Also explain losses taking place in cooling tower. **04**
- (c) In detail explain about Pressure Swing Adsorption. Also cite few industrial examples of the same. **07**
