| Seat No.: | Enrolment No. |
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GUJARAT TECHNOLOGICAL UNIVERSITY

BE - SEMESTER-VII (NEW) EXAMINATION – WINTER 2022

| Subj | ect | Code:3170511 Date:12-02 | 1-2023 |
|-----------------|-------------------|---|-------------------------|
| _ | | Name:Transport Phenomena :30 AM TO 01:00 PM Total Mai | rks:70 |
| Instru | | | 115070 |
| | 2. 3. | Attempt all questions. Make suitable assumptions wherever necessary. Figures to the right indicate full marks. | |
| | 4. | Simple and non-programmable scientific calculators are allowed. | |
| Q.1 | (a) (b) (c) | Discuss the three levels at which the transport phenomena is studied. | Marks 03 04 07 |
| Q.2 | (a) | Define Prandtl number. | 03 |
| Q.2 | (b) | | 04 |
| | (c) | | 07 |
| | | OR | |
| | (c) | Derive equation for momentum for a Newtonian incompressible fluid. | 07 |
| Q.3 | (a) | Write general shell momentum balance equation. | 03 |
| C | (b) | • | 04 |
| | (c) | | 07 |
| Q.3 | (a) (b) | Define Reynold's number. A fluid of viscosity 0.7 cP is present between two parallel plates separated by a distance 0.001 ft. If the lower plate moves with the velocity 1ft/sec in positive x direction, compute the steady state momentum flux τ_{yx} in | 03 04 |
| | (c) | Pascal. Derive the equation for pressure drop for laminar flow of an Newtonian, incompressible fluid through a circular cross section pipe. | 07 |
| Q.4 | (a) | Define the molecular heat flux. | 03 |
| Q. - | (b) | | 04 |
| | (c) | Derive equation for temperature drop in a conductor wire for an electrical heat source The rate of heat production per unit volume is given by the expression $S_e = I^2/k_e$. | 07 |
| Q.4 | (a) | OR State the Fourier's law of heat conduction. | 03 |
| Ų. 1 | (a) (b) | | 03 |
| | (c) | Derive the equation for temperature distribution for chemical reaction as | 07 |

heat source.

| Q.5 | (a) | Define Mass and Molar Concentrations, Mass Average and Molar | 03 |
|-----|------------|---|----|
| | | Average Velocities, Molecular Mass and Molar Fluxes | |
| | (b) | Explain temperature and pressure dependency of diffusivity. | 04 |
| | (c) | Derive the relation for diffusivity in case of equimolar counter diffusion. | 07 |
| | | OR | |
| Q.5 | (a) | Define binary Diffusivity. | 03 |
| | (b) | Shortly explain the theory of diffusion in gases at low density. | 04 |
| | (c) | With neat diagram derive an equation of molar flux for the diffusion with | 07 |
| | | heterogeneous chemical reaction as $2A \rightarrow B$. | |
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